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CIA-RDP86-00513R001963310015-2

YUZVUK, V.YE.

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ENGINEER

APPROVED FOR RELEASE: 03/15/2001

CIA-RDP86-00513R001963310015-2"

PAUSHKIN, Ya.M., YUZVYAK, A.G.

Synthesis of vinylcyclohexane and styrene from butadiene. Neftekhimiia
1 no.1:60-64 Ja-F '61. (MIRA 15:2)

1. Institut neftekhimicheskoy i gazovoy promyshlennosti imeni
M.I.Oubkina.
(Styrene) (Butadiene) (Hexane)

g/152/61/000/002/002/005
B124/B203

AUTHORS: Paushkin, Ya. M., Yuzvyak, A. G.

TITLE: Cyclopolymerization of butadiene with production of vinyl cyclohexene

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy. Neft' i gaz, no. 2,
1961, 69-74

TEXT: The authors studied the cyclopolymerization of butadiene and the further chemical conversion of the dimer. In the thermal polymerization at 400-500°C, the reaction may proceed to the dimer or trimer. The dimer yield attained by S. V. Lebedev was 85-86% at 150°C after 5 days. A dimer yield of about 80% was attained in an experiment in an enameled bomb at 150°C after 120 hr (S. V. Lebedev and S. R. Sergeyenko (Ref. 4)). Vinyl cyclohexane is produced in the hydrogenation of vinyl cyclohexene, and vinyl decalin in the hydrogenation of vinyl decalene, whereas styrene and vinyl naphthalene are produced in the dehydrogenation of the compounds mentioned. The cyclopolymerization was conducted in a flow reactor made of quartz glass with a central canal for the thermocouple (Fig. 1). The reactor tube was filled

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S/152/61/000/002/002/005
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Cyclopolymerization of ...

with the charge to be investigated or with the catalyst. The liquid polymer was collected in a receptacle, and the gas in a gasometer. The polymer was distilled in a laboratory column, the fraction obtained at 128-133°C, and its physicochemical constants were determined. The following catalysts were used: phosphoric acid on kieselguhr, and chromium oxide on aluminum oxide. Experiments without a catalyst were also made, first with glass packing, then with activated carbon. The effect of temperature on yield and properties of polydivinyl at constant volume velocity was studied. At constant volume velocity, the liquid-polymer yield as well as the specific gravity and the refractive index increase. The amount of unsaturated compounds in the liquid polymer drops with rising temperature, whereas in the presence of the fraction 128-133°C, which also contains the vinyl cyclohexene, it rises with temperature, and drops after reaching a certain maximum (Fig. 3). The polymer yield increases both with respect to the initial butadiene and to the polymer with the volume velocity of the supply of raw material at constant temperature (400°C) (Figs. 4, 5); a supplying rate of the raw material of 12 h^{-1} is optimum. The dimerization was conducted under equal conditions (temperature, volume velocity) on activated carbon

Card 2/7

S/152/61/000/002/002/005
B124/B203

Cyclopolymerization of ...

and glass packing; the yield of the fraction 128-133°C was higher with activated carbon; the same applies to the polymer yield (Fig. 6). Butadiene polymerized in the presence of H_3PO_4 on kieselguhr only at 400°C, the polymer yield after one passage being 39.4% of the initial butadiene, and the dimer yield 8% of the liquid polymer. At 450°C, the polymer yield was 31.6%. The formation of all three xylenes can be assumed on the basis of the specific gravity, the refractive index at 20°C, and the aniline point. In the presence of a Ni-Cr catalyst, the degree of conversion was 13.2% at 400°C, and 17.7% at 490°C. The product mainly consisted of aromatic hydrocarbons (xylenes). The fraction 128-133°C distilled in a laboratory column delivered almost pure vinyl cyclohexene (up to 95% yield). The resulting cyclohexene was selectively hydrogenated (Ref. 6) on a catalyst (10% Pt on activated carbon); here, a product was obtained whose constants corresponded perfectly to those of vinyl cyclohexane. There are 6 figures, 2 tables, and 6 references: 4 Soviet-bloc and 2 non-Soviet-bloc.

ASSOCIATION: Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti im. akad. I. M. Gubkina (Moscow Institute of the Petrochemical and Gas Industry imeni Academician I. M. Gubkin)

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S/152/61/000/002/002/005
B124/B203

Cyclopolymerization of ...

SUBMITTED: December 30, 1960

Legend to Fig. 1: Diagram of the laboratory apparatus: 1) gasometer,
2) rheometer, 3) CaCl_2 tube, 4) electric furnace,
5) transformer, 6) millivoltmeter, 7) cooler,
8) receptacle, 9) trap, 10) quartz tube.

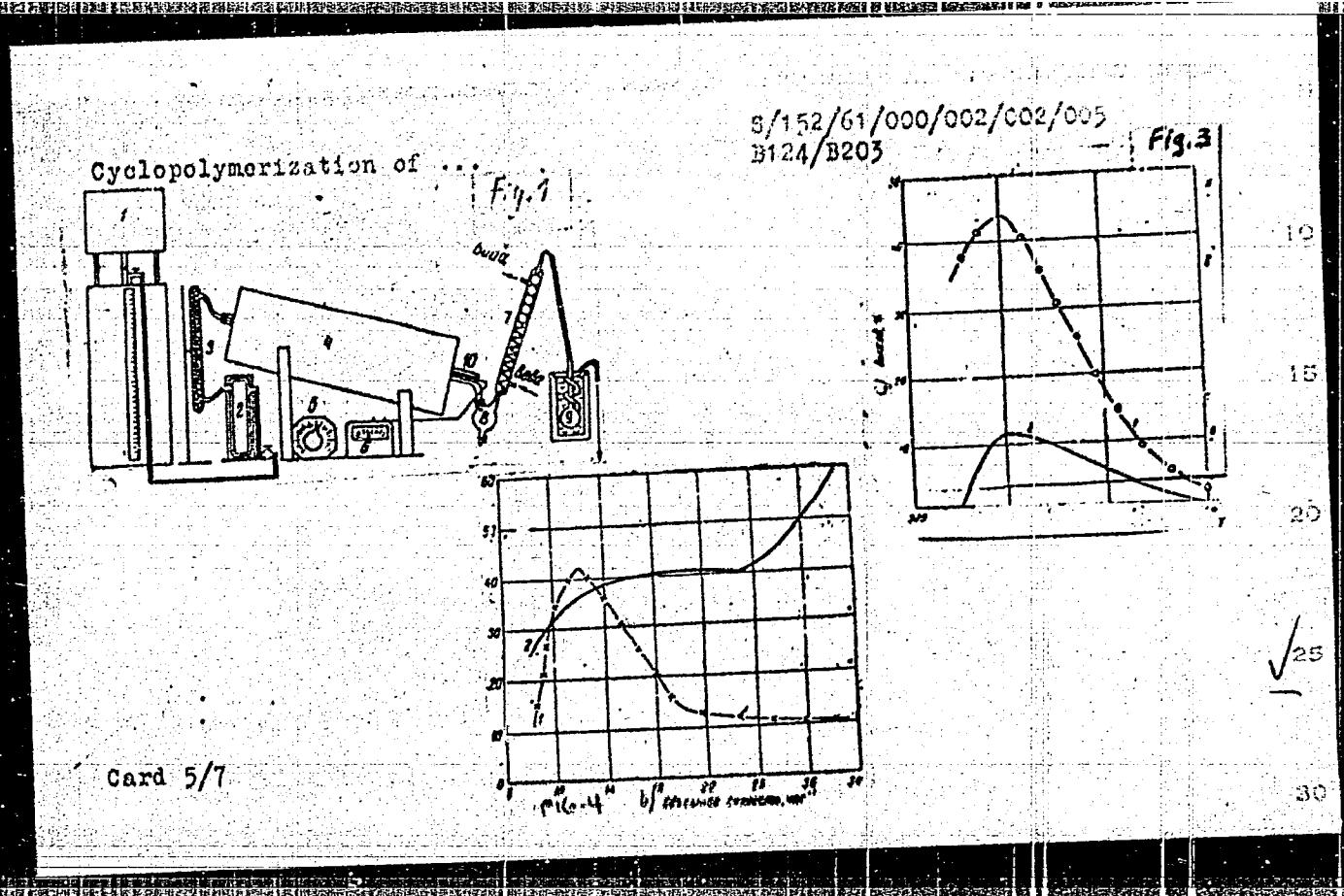
a) water, b) to the gasometer

Legend to Fig. 3: Effect of temperature on the yield
of fraction 128-133°C at constant volume velocity:

1) yield of the fraction referred to the liquid polymer,
2) yield of the fraction referred to the initial
butadiene; a) yield, %, b) temperature, °C

Legend to Fig. 4: Effect of the volume velocity on the
yield of fraction 128-133°C at 400°C: 1) referred to
divinyl, 2) referred to the polymer; a) yield, %,
b) volume velocity, cm^3/min

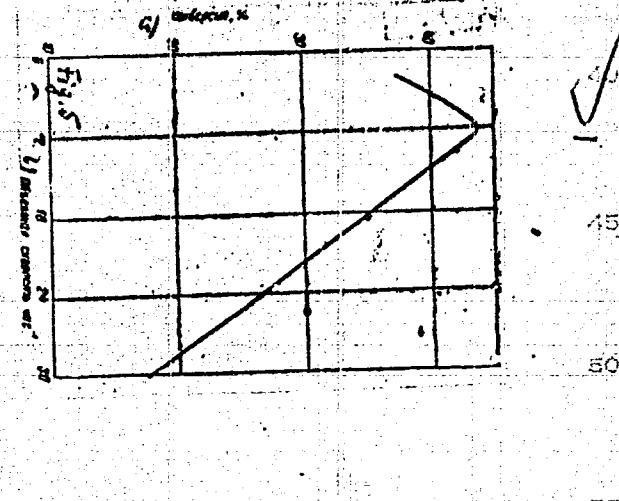
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Cyclopolymerization of ...

S/152/61/000/002/002/005
B124/B203

Legend to Fig. 5: Effect of the volume velocity on the conversion at 400°C referred to the initial butadiene,
a) conversion, %, b) volume velocity, $\text{cm}^3 \text{min}^{-1}$

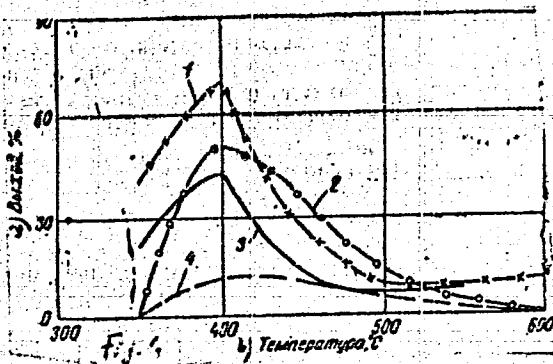


Card 6/7

S/152/61/000/002/002/005
B124/B203

Cyclopolymerization of ...

Legend to Fig. 6: Effect of the packing on the yield of fraction 128-133°C referred to the reacted butadiene: 1) with activated carbon, 2) with glass packing referred to the initial butadiene, 3) with activated carbon, 4) with glass packing; a) yield, b) temperature



Card 7/7

YUDEVYAK, A.G.; PAUSHKIN, Ya.M.

Trends in the dehydrogenation reaction of vinylcyclohexene. Izv.
vys.ucheb.zav.; neft' i gaz 5 no.8:85-89 '62. (MIRA 17:3)

1. Moskovskiy institut neftekhimicheskoy i gazovoy promyshlennosti im.
akademika I.M.Gubkina.

S/020/62/144/003/023/030
B124/B101

AUTHORS:

Paushkin, Ya. M., Yuzvyak, A. G., and Rubinshteyn, A. T.

TITLE:

Synthesis of dimethyl cyclohexadiene and vinyl cyclohexene
by dimerization of butadiene

PERIODICAL:

Akademiya nauk SSSR. Doklady, v. 144, no. 3, 1962, 581-584

TEXT: Optimum conditions for the cyclopolymerization of butadiene to cyclic dimers of various compositions were studied in a stainless-steel reactor with activated-carbon packing. The polymer obtained was subjected to fractional distillation, and cuts with boiling-point intervals from 5 to 7°C were collected and examined. Maximum yields of dimeric fractions were obtained at 400-420°C, with a feeding velocity of 11 hrs⁻¹, and 3 atm pressure, corresponding to 10% 1,3-dimethyl cyclohexadiene and 45% vinyl cyclohexene. The yield of dimers decreases with increasing reaction temperature, and increases with increasing pressure. In addition to the dimethyl cyclohexadienes and vinyl cyclohexene, long-chain aromatic compounds and both cyclooctadiene- and cyclodecene-type hydrocarbons were shown to be present. The thermodynamics of the reactions:

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Synthesis of dimethyl cyclohexadiene ...

S/020/62/144/003/023/030
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$\xrightleftharpoons[K_p]{K'_p}$ divinyl vinyl cyclohexene (I); $\xrightleftharpoons[K_p]{K'_p}$ divinyl vinyl cyclohexene (I),
 $\xrightleftharpoons[K_p]{K'_p}$ dimethyl cyclohexadiene (II) were calculated in the gas phase from
the equations (1) $K_p = p_{vin}/p_{div}^2$ for reaction I, and both (2)
 $K_p^{(1)} = p_{vin}/p_{div}^2$ and (3) $K_p^{(2)} = p_{dimethyl}/p_{div}^2$ for reaction (II), where
 K_p , $K_p^{(1)}$ and $K_p^{(2)}$ are equilibrium constants of the two reactions at con-
stant pressure, and p_{vin} , p_{div} , and $p_{dimethyl}$ are the equilibrium partial
pressures of vinyl cyclohexene, divinyl, and dimethyl cyclohexadiene,
respectively. If $z \cdot 100$ is the percentage of vinyl cyclohexadiene in reac-
tion, I, $x \cdot 100$ that in reaction II, $y \cdot 100$ the percentage of dimethyl
cyclohexadiene in reaction II, and if P_0 is the pressure required, we have
(1") $K_p = (1/2)z[1 - (1/2)z] / (1-z)^2 \cdot P_0$ for reaction I,

$$(2'') K_p^{(1)} = x[1 - (1/2)x - (1/2)y] / (1 - x - y)^2 \cdot P_0 \text{ and}$$

$$(3'') K_p^{(2)} = y[1 - (1/2)x - (1/2)y] / (1 - x - y)^2 \cdot P_0 \text{ for reaction II. At}$$

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Synthesis of dimethyl cyclohexadiene...

S/020/62/144/003/023/030
B124/B101

the same time, $-RT \ln K_p = \Delta Z^\circ$; $\Delta Z^\circ = \Delta H_{298.16}^\circ - T\Delta S_{298.16}^\circ$;
 $\Delta H_{298.16}^\circ = \Delta H_{\text{form.}298.16C_8}^\circ - 2\Delta H_{\text{form.}298.16C_4}^\circ$; $\Delta S_{298.16}^\circ = S_{298.16C_8}^\circ - 2S_{298.16C_4}^\circ$, where ΔZ° is the thermodynamic potential, $\Delta H_{298.16C_4}^\circ$ = heat of formation of divinyl, $S_{298.16C_4}^\circ$ the entropy of divinyl, and $S_{298.16C_8}^\circ$ = entropy of the corresponding dimer. There are 3 figures and 3 tables. The English-language reference is: E. Gil. Av. J. Shabtal. F. Steckel, Ind. Eng. Chem. 52, 31 (1960). [Abstracter's note: p.139 given in the original, is wrong.]

ASSOCIATION: Institut neftekhimicheskoy i gazovoy promyshlennosti im. I. M. Gubkina (Institute of Petrochemical and Gas Industry imeni I. M. Gubkin)

PRESENTED: December 30, 1961, by A. V. Topchiyev, Academician

SUBMITTED: December 30, 1961

Card 3/3

ACCESSION NR: AT4008696

8/2982/63/000/044/0027/0033

AUTHOR: Paushkin, Ya. M.; Yur'yak, A. G.

TITLE: Synthesis of new monomers from (1,3) -butadiene

SOURCE: Moscow. Institut neftekhimicheskoy i gazovoy promyshlennosti. Trudy*, no. 44, 1963. Neftekhimiya, pererabotka nefti i gaza, 27-32

TOPIC TAGS: 1,3-butadiene, 1,3-butadiene polymerization, 1,3-butadiene dimerization, cyclohexene, vinyl-, cyclohexane, vinyl-, 1,3-butadiene thermal dimerization, monomer

ABSTRACT: The authors first studied the thermal cyclodimerization of 1,3-butadiene to vinylcyclohexene and 1,3-dimethylcyclohexadiene over activated charcoal at 350-500C and 2-3 atmospheres. After dimerization, the liquid polymer was fractionated and the fractions boiling at 124-128 and 128-132C were collected. It was found that the yield of total polymer increases with the temperature, but that the yield of dimer decreases, so that the optimal temperature is 400C. At this temperature, the yield increases with a decrease in the rate of flow of the monomer. The authors then studied the selective catalytic hydrogenation of vinylcyclohexene over Pt at room temperature, yielding vinylcyclohexane, as well as its chlorination with Cl₂ in CCl₄ at -60C, yielding vinylchlorocyclohexane, and its

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ACCESSION NR: AT4008696

hydrochlorination with HCl in the presence of anhydrous $SnCl_4$ in CCl_4 at -65C, yielding vinylchlorocyclohexane, chloroethylcyclohexene and vinylcyclohexene dihydrochloride; the latter reaction did not take place in any solvent in the absence of a catalyst and was not catalyzed by $TiCl_4$, $ZnCl_2$ or $FeCl_3$. Orig. art. has: 3 figures, 5 tables and 2 structural formulas.

ASSOCIATION: Institut neftekhimicheskoy i gazovoy promyshlennosti, Moscow (Institute of Petroleum Chemistry and the Gas Industry)

SUBMITTED: 00

DATE ACQ: 16Jan64

ENCL: 00

SUB CODE: QC,PP

NO REF Sov: 007

OTHER: 007

2/2

Card

PAUSHKIN, Ye.M.; YUZYAK, A.G.

Obtaining new monomers on a base of butadiene. Trudy MINSKIP
(MIRA 36:5)
no.44:27-33 '63.

VYTAUTAS, K.
VYTAUTAS, K., inzh.; KREGZDAILE, D., red.; LIEGUS, S., tekhn. red.

[New techniques in the building materials industry of the
Lithuanian S.S.R.] Naujoji technika Lietuvos TSR statyviniu
medziagu pramoneje. Vilnius, Centrinis techninės informacijos
ir propagandos biuras, 1960. 126 p. [In Lithuanian]
(MIRA 14:12)

(Lithuania--Building materials industry)

YODELYTE
DEGUTIS, Yu.; YODELYTE, A. [Jodelyte, A.]

Synthesis of β -{ α -[bis(dichloroethyl)amino]phenyl}-dl- β -alanine. Zhur. obshch. 32 no. 2:567-570 F. 62. (MIRA 15:2)

1. Vil'nyuskiy gosudarstvennyy universitet imeni V. Kapsukasa.
(Alanine)

YYGI, A. Yu. [Jõgi, A.J.]

History of bird banding in the Estonian S.S.R. Migr. zhiv. no.1:40
47 '59. (MIRA 13:6)

1. Institut zoologii i botaniki AN Estonskoy SSR.
(Estonia-Bird banding)

L 01196-66 EWT(l)/EWP(e)/EWT(m)/EWP(i)/EWP(t)/EWP(b) IJP(c) MI WH

ACCESSION NR: AT 013690

UR/2613/64/000/030/0051/0067

AUTHOR: Yygi, Kh. R.

TITLE: On a procedure and results of comparison of the values of adhesion and cohesion of certain sublimated layers

SOURCE: AN EstSSR. Institut fiziki i astronomii. Trudy, no. 30, 1964. Issledovaniya po luminescencii (Research on luminescence), 51-67

TOPIC TAGS: thin film, adhesion force, cohesion force, sublimated film, alkali halide film, sublimated luminescence

ABSTRACT: The purpose of the investigation was a more detailed study of the physical properties arising during the preparation of sublimated phosphor, a procedure for the production of which was first developed by F. D. Klement (Op. i spektr. v. 1, 571, 1956). The experiments were made on non-activated alkali-halide thin films deposited on substrates of fused quartz and LK-5 glass, sublimated from single-crystal KCl and NaCl. A study was made of the dependence of the adhesion and cohesion of these layers (thickness 1--4.5 nm) on the temperature of the substrates during the time of deposition of the layer, in the temperature interval 33--225°C. The apparatus used for the sublimation and for the measurement of adhesion forces are described, and the instrument for the determination of adhesion is shown in

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L 01196-66

ACCESSION NR: AT5013690

Fig. 1 of the enclosure. The results showed that the adhesion layer decreased by approximately 5% as the substrate temperature increased from 35 to 85°C, while the cohesion in the layer remained unchanged. In NaCl the cohesion forces are approximately double those of KCl layers. "I am deeply grateful to P. D. Clement for suggesting the topic and guidance, and to V. Malyshcheva for a discussion of the results." Orig. art. has: 4 figures.

17.55

6

ASSOCIATION: none

SUBMITTED: 13Oct84

ENCL: 01

SUB CODE: OP

NR REF Sov: 003

OTHER: 013

Card 2/3

01196-65
ACCESSION NR: A15013690

INCLUSION: 01

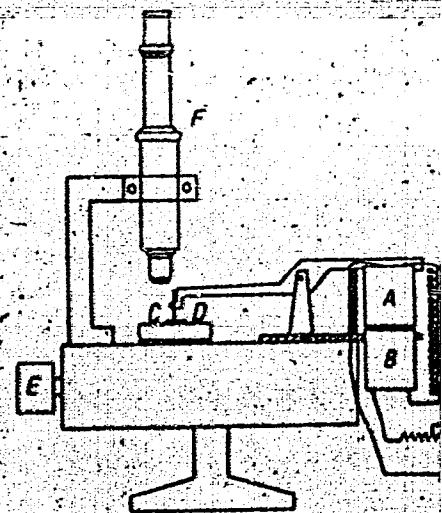


Fig. 1. Diagram of instrument for the determination of adhesion. A, B - dc magnets; C - needle; II - investigated object on stand; E - motor to move stand; M - microscope

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YYGI, Kh. R.

26390
S/613/61/000/017/008/011
DC51/D113

243500 (1137, 1138, 1163)

AUTHORS: Malysheva, A.F. and Yogi, H.R.

TITLE: Preparation and investigation of zinc sulfide sublimate phosphors

SOURCE: Akademiya nauk Estonskoy SSR. Institut fiziki i astronomii. Trudy, no. 17, 1961. Issledovaniya po luminesentsii, 109-119

TEXT: ZnS-Mn, ZnS-Cu and ZnS-Zn sublimate phosphors with intense emission upon cathode and photo excitation were prepared and their absorption, excitation, and irradiation spectra measured. The known methods of preparing phosphors by sublimation are summarized and the special procedures used by the authors described. By comparing the stationary luminescence and phosphorescence excitation spectra of these phosphors with their absorption spectra, and also by comparing luminescence intensity growth curves for excitation in different regions of the absorption spectrum, the possibility is discussed of regarding the $\lambda_{\text{max}} = 218 \text{ m}\mu$ band as an exciton band and

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Preparation and investigation ...

S/613/61/000/017/008/011
D051/D113

the edge of absorption in the 325-330 μm region as being due to band-to-band transition. For more reliably identifying the bands in the ZnS absorption spectrum, a study of the dependence of photoconductivity on the wavelength of the incident light and also investigations of infrared effects and other physical processes during excitation of luminescence in different absorption bands are recommended. F.D.Klement, Ch.B.Lushchik, and K.-S.K. Rebane are thanked for help rendered. There are 10 figures.

SUBMITTED: April 29, 1961

Card 2/2

YYGIS, V.A.

New nematodes of birds from the Courland Lagcon. Trudy Sonl.
inst. 35:208-215 '65. (MIRA 19:1)

1. Zoologicheskiy institut AN SSSR.

YYGISTE, A.K. [Jogiste, A.]; GUSEV, V.M.

Toxoplasmosis in wild birds of the U.S.S.R. Dokl. AN SSSR
143 no.2:491-492 Mr '62. (MIRA 15:3)

1. Institut epidemiologii i mikrobiologii im. N.F.Gamaliya
AN SSSR i Nauchno-issledovatel'skiy protivchumnyy institut
Kavkaza i Zakavkaz'ya. Predstavлено akademikom Ye.N.Pavlovskim.
(CAUCASUS—TOXOPLASMOSIS)
(PARASITES—BIRDS)

YYOISTH-A.K. [Jogiste, A.]

Penetration of toxoplasma into the eggs of infected hens. Dokl.
AN SSSR 148 no.4:989-990 F '63. (MIRA 1614)

1. Institut epidemiologii i mikrobiologii im. N.F.Gamaleya
AMN SSSR. Predstavлено академиком Ye.N.Pavlovskim.
(Toxoplasmosis) (Eggs—Microbiology)

ZASUKHIN, D.N.; YYGISTE, A.K.

Something new in the study of toxoplasmosis. Trudy Inst.
zool. AN Azerb. SSR 23:125-133 '64. (MIRA 17:9)

OIGLANE, H.

Integration in four-dimensional pseudo-Euclidean space.

P. 85, (Uurimused Trudy) No. 5, 1957, Tallinn, Estonia

SO: Monthly Index of East European Acessions (EEAI) Vol. 6, No. 11 November 1957

OIGLANE, H.

A relativistically invariant treatment of the self-action of a free electron in quantum electrodynamics.

P. 93, (Vurimused Trudy) No. 5, 1957, Tallinn, Estonia

SO: Monthly Index of East European Accessions (EEAI) Vol. 6, No. 11 November 1957

OIGLANE, H.

A treatment of the self-action of a bound electron in quantum electrodynamics.

P. 100, (Uurimused Trudy) No. 5, 1957, Tallinn, Estonia

SO: Monthly Index of East European Accessions (EEAI) Vol. 6, No. 11 November 1957

Yyglane, Kh.

56-6-32/47

AUTHOR: Yyglane, Kh.

TITLE: A Wave Equation for the Free Nucleon (Volnovoye uravneniye dlya svobodnogo nuklona)

PERIODICAL: Zhurnal Eksperimental'noy i Teoreticheskoy Fiziki, 1957, Vol. 33, Nr 6 (12), pp. 1511 - 1512 (USSR)

ABSTRACT: It has hitherto not been possible to set up a wave equation for the nucleon (not even for the free nucleon). This is partly due to the fact that the mass in proton states and neutron states differs. In this respect it is interesting that the linearization of the wave equation of second order in the general case leads to the equation $\left[\gamma^5 \partial/\partial x - (m_0 c/\hbar) \exp(a \gamma^5) \right] \Psi = 0$. Here m_0 and a denote certain constants. The author then generalizes the above equation as follows: $\left[\Gamma^5 \partial/\partial x - (m_0 c/\hbar) I \exp(2a T_3) \right] \Psi = 0$. Here the commutation relations for the operators Γ^5 , I and T_3 are determined by the following conclusions: a) The Hamiltonian, the operator of the projection of the spin, and the third component of the isotopic spin, T_3 , form a commutating system of operators. b) The relation between energy and momentum has the usual form. Herefrom

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56-6-32/47

A Wave Equation for the Free Nucleon

follows: $[\Gamma^Y \Gamma^\sigma]_+ = 2\delta_{\sigma\sigma}$, $[\Gamma^Y T_3]_+ = 0$, $[T_3, I]_+ = 0$,

$[\Gamma^Y I]_- = 0$, $T_3 T_3 = 1$, $I I = 1$. For the operators Γ^Y , T_k and I there exists eight-rowed irreducible representations:

$$\Gamma^Y = 1^{II} \times Y^Y, T_k = \sigma_k \times Y^5, I = \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix} \times 1^{IV}$$

A table contains those quantum numbers which, in the relativistic limit, characterize the components of the wave function. The eigenvalue $t_3 = -1/2$ corresponds to the particle mass $m_1 = m_e^{-8}$, and the eigenvalue $t_3 = 1/2$ corresponds to the mass $m_2 = m_e^{0a}$. If one puts $m_e = 1837,38 m_e$ (average value of the masses of the protons and neutrons) and $a_e = 3\alpha/10\pi$, then m_1 and m_2 agree within the limits of error with the experimental masses of the proton and of the neutron. The second equation mentioned here is then the wave equation of the free nucleon. The expressions for the specific nucleon current, the electric current, and the operator of the electric charge of the nucleon, following from this equation, are also given. This wave equation of the nucleon is not invariant with respect to not real Lorentz transformations, but it is invariant with respect to simultaneous reflections of spatial and isotopic coordinates. By means of the wave equation given here it is possible

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A Wave Equation for the Free Nucleon

56-6-32/47

to describe also the doublet of Ξ -hyperons. There are 1 table
and 1 Slavic reference.

ASSOCIATION: Institute for Physics and Astronomy AN Estonian SSR
(Institut fiziki i astronomii Akademii nauk Estonijskoy SSR)

SUBMITTED: July 15, 1957

AVAILABLE: Library of Congress

Card 3/3

Yyglane, Kh.

56-6-47/47

AUTHOR: Yyglane, Kh.

TITLE: A Possibility for the Classification of Barions (Odna vossobraznost'
dlya klassifikatsii barionov)

PERIODICAL: Zhurnal Eksperimental'noy i Teoreticheskoy Fiziki, 1957, Vol. 33,
Nr 6, pp. 1537-1538 (USSR)

ABSTRACT: From the wave equation for barions the general form holds good for the
electric charge of the barions: $q = -\epsilon(t_3 + v_3)$. t_3 is the eigen-
value of the operator T_3 . v_3 is a number which changes in transition
from one multiplet to another. In the investigation carried out it re-
places the Gell-Mann number and can be understood to be the eigenvalue
of the third component of the operator for the isotopic moment V_3 . The
postulate that a half-integer or a multiple of a half-integer charge
must be lacking in elementary particles results in the rule:
a) t_3 and v_3 are at the same time half-integer and integer,
b) $|t_3 + v_3|$ can never exceed 1.

Herefrom follows the quantum number which determines the nature of
the barions:

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A Possibility for the Classification of Barions

56-6-47/47

Nature of barions

 t_3 v_3

P	- 1/2}	- 1/2
n	+ 1/2}	
Λ^0	0	0
Ξ^-	- 1/2}	+ 1/2
Σ^-	+ 1/2}	
Σ^0	1	
Σ^+	0}	0
	- 1	

It may be seen herefrom that the nucleons, Ξ^- -hyperons, and Λ^0 -particles may in reality be assumed to be "similar" particles, whereas the Σ hyperons form a group of their own. There are 1 table and 3 references, 1 of which is Slavio.

ASSOCIATION: Institute of Physics and Astronomy AN Estonian SSR (Institut fiziki i astronomii Akademii nauk Estonskoy SSR)
 SUBMITTED: July 15, 1957
 AVAILABLE: Library of Congress
 Card 2/2

AUTHOR:

Yyglane, Kh.

SOV/56-34-5-49/61

TITLE:

On a Systematic Classification of Mesons and Baryons (K sistematicheskaya klassifikatsiya mesonov i barionov).

PERIODICAL:Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1958,
Vol. 34, Nr 5, pp. 1337 - 1338 (USSR)**ABSTRACT:**

In a previous paper (Ref 1) the author gave a classification of the baryons on the basis of two quantum numbers, namely of the third component of the isotope spin t_3 and of the third component of the so-called isotope moment v_3 . These results from this paper are compiled in a table. To obtain an analytical scheme for the mesons the irreducible equations for the multiplets of the free bosons must be determined. Besides, an ordinary wave equation of second order results from the equation

$\beta \partial/\partial x + k \exp(a_5) = 0$. β and k denote the matrices by Kemmer-Duffin (Deffin), a denotes a certain constant.

Card 1/4

On a Systematic Classification of Mesons and Baryons SOV/56-34-5-49/61

5 1 2 3 4 holds. The initially given equation is a generalisation of the equation by Proca-Kemmer-Duffin (Deffin). This equation, written down at the beginning, however, is not appropriate for the description of the multiplets of the particles. The author generalizes the above equation by

$B \frac{\partial}{\partial x} + k_0 I \exp(\alpha T_3) = 0$. Here the operators B_σ , I , and T_3 satisfy the exchange relations $B_\sigma B_\sigma + B_\sigma B_\sigma = \delta_{\sigma\sigma}$, $B_\sigma + \delta_{\sigma\sigma} B_\sigma$, $B_\sigma T_3 + T_3 B_\sigma = 0$, $IT_3 + T_3 I = 0$, $IB_\sigma - B_\sigma I = 0$. The irreducible representation

$B_\sigma = 1^{II} \times \beta_3$, $T_3 = \sigma_3 \times \tau_5$, $I = \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix} \times 1^V$ can be chosen for the operators. In the case of such a choice the aforementioned equation describes the multiplets of the free mesons with regard to the isotopic spin. The operator of the isotope moment v_3 can be introduced in a perfectly analogous way as in the case of the fermions. The corresponding quantum number for the characterisation of the mesons is given in a table. The system of the

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On a Systematic Classification of Mesons and Baryons SOV/56-34-5-49/61

mesons agrees with the system of the baryons. The electrical charge of the baryon and of the meson is expressed by the common formula $q = -e(t_3 + v_3)$. The study of the experimental material delivers the following rules: a) In the case of strong and of electromagnetic interaction the third component of the isotope spin as well as the third component of the isotope moment of the system are maintained. The electrical charge, of course, is maintained as well. b) In the case of weak interaction only the charge of the system is maintained because then the third component of the isotope spin and the third component of the isotope moment change by $\pm 1/2$. All processes which do not satisfy these rules are forbidden. There are 2 tables and 4 references. which are Soviet.

ASSOCIATION: Institut fiziki i astronomii Akademii nauk Estonskoy SSR
(Institute of Physics and Astronomy of the Estonian SSR)

SUBMITTED: January 31, 1958
Card 3/4

On a Systematic Classification of Mesons and Baryons Sov/56-34-5-49/61

1. Mesons--Classifications 2. Mesons--Analysis

Card 4/4

X Y G L A N E, K.

2(1.8)) 2(5) PLATE I BOOK EXPLORATION - 207/359

Yazykovnaya elementarnykh chastits po krotoem svedenii
1 tsvet. Moscow, 1958.
Problemy sovremennoy teorii elementarnykh chastits. No. 2: Teoriya
konfrentatsii... [Problems in the Modern Theory of Elementary
Particles. No. 2: Transactions of the All-Union Inter-Union
Conference on the Quantum Field Theory and the Theory of
Elementary Particles] Uchgorod, Zakarpatskoye oblastnoye izdat-
stvo, 1959, 21, p. 5,000 copies printed.

M.: Yu. Lomidze, Docent; Tech. Ed.: N. M. Balonc.

PURPOSE: This book is intended for physicists, particularly those
concerned with problems in the field of elementary particles and
the quantum theory.

CONTENTS: This book contains articles on elementary particles
originally read at the All-Union Inter-Union Conference held at
Uzhgorod State University on October 26, 1958. Among the topics
discussed are: the spinor field theory, the fusion theory,
Lorentz contractions, parity studies, nucleon-nucleon scattering,
etc. English abstracts accompany each article. References
follow each article.

Sobolik, G.A.	New Formulation of Pauli Theory	2
Samokhin, Yu.N. and B.I. Rukhman	Application of Schrödinger's Variation Method to the Pauli Theory	20
Sobolik, G.A.	Generalization of the Lorentz Group	27
Frantsuzova, O.J.	Generalized Equivalent Potential and the Sequence of Infraregional Lorentz Contractions Under Rotary Motion	41
Sobolik, G.A.	Representation of the Complete Lorentz Group	52
Sobolik, G.A.	Connection Between the "Aneural" Representation of the Space-Time Inversion Group and the Pauli Transformations	58
Shestopalov, V.Ya., S.A. Moshnikov, and A.P. Butik	Non-conservation of Parity in Had Particle Processes	58
Sobolik, G.A. and A.M. Syudin	Determination of Parity for Strange Particles	63
Lomidze, Yu.M.	The Possible Variations of the β - Decay Theory	63
Smirnov, V.S.	Violation of the CPT Order for Spinor Wave Functions	63
Sobolik, V.Z.	Conservation of the Chiral Parity as a Fundamental Law of the Symmetry in Nature	63
Geshkenbein, B.V.	Polarization of Electrons or the Nuclear Contraction Due to β^- Decay. Taking Into Account the Electric Field of the Nucleus	69
Zil'blat, A.G.	Polarization of the Nucleons Under the Strong-Force Interaction in the High Energy Region	71
Bogolyubov, N.N.	Wave Equations for Elementary Particles	109
Sarkisyan, V.S. and B.M. Bakhshayev	Some Remarks on the Inner Structure of the Nucleon	117
Sobol'zov, V.G.	On the Superfluid State of an Atom Nucleus	125

21(1)

AUTHOR:

Yuglone, Kh. (Oiglane, H.)

SOV/56-37-2-38/56

TITLE:

A Comment on the Scheme of Baryons

PERIODICAL:

Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1959,
Vol 37, Nr 2(8), pp 558-559 (USSR)

ABSTRACT:

The author assumes that each of the eight baryons can be described by a four-component wave function. The common equation for all baryons is in this case an equation for a 32-component spinor. The 32-component spinor space can be treated as a mapping of a 10-component vector space. The unit vectors of this space correspond to 32-row matrices Γ^a , where $|\Gamma^a|^2 + |\Gamma^b|^2 = 2\delta_{ab}$. The following transformation of the matrices $\Gamma^a \rightarrow \Gamma^{a'} = -\Gamma^b \Gamma^a \Gamma^b$, $a = 1, 2, \dots, 10$ (no summation over b) corresponds to the reflection of the unit vectors e^b in the vector space. The author subdivides the 10-dimensional vector space into a fourdimensional Minkowski-space (the matrices corresponding to the unit vectors are Γ^j , $j = 1, 2, 3, 4$) and into a fourdimensional

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A Comment on the Scheme of Baryons

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sional isotopic space (matrices Γ^a , $a = 5, 6, \dots, 10$). By means of transformations of the isotopic space the following three-component isotopic vectors can be defined (their components corresponding to the same relationships as the components of ordinary spin: $2\vec{J}^1 = (\Gamma^6, \Gamma^5, i\Gamma^5\Gamma^6)$, $2\vec{J}^2 = (\Gamma^8, \Gamma^7, i\Gamma^7\Gamma^8)$, $2\vec{J}^3 = (\Gamma^{10}, \Gamma^9, i\Gamma^9\Gamma^{10})$, $2\vec{J}^4 = (i\Gamma^6\Gamma^7, i\Gamma^7\Gamma^5, i\Gamma^5\Gamma^6)$, $2\vec{J}^5 = (i\Gamma^{10}\Gamma^8, i\Gamma^8\Gamma^9, i\Gamma^9\Gamma^{10})$). The common equation for all components should, say, have the form: $\left\{ \Gamma^\nu \frac{\partial}{\partial x_\nu} - k_0 I \exp \left[\Gamma^0 \sum_{k=1}^5 a_k J^k \right] \right\} \psi = 0$, where $\Gamma^0 = \Gamma^1\Gamma^2\Gamma^3\Gamma^4$, $I = i\Gamma^0\Gamma^6\Gamma^7\Gamma^{10}$, the a_k denoting certain small isotopic vectors, the components of which depend upon the potentials of the fields of the pions and leptons. The last equation is solved by perturbational methods, the Hamiltonian of the undisturbed problem being chosen as $H = E_c \left\{ \Gamma^4 \Gamma^k \frac{\partial}{\partial x_k} - \right.$

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A Comment on the Scheme of Baryons

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$- k_0 \Gamma^4 I \exp \left[\Gamma^0 \sum_{k=1}^5 a_k \frac{\Gamma^k}{\Gamma^{k+3}} \right] \right\}$. The form of the baryon is characterized by three quantum numbers τ_3, ω_3, f_3 , which are the eigenvalues of the operators $(1/2) \Gamma^0 \Gamma^7 \Gamma^8$, $(1/2) \Gamma^0 \Gamma^9 \Gamma^{10}$, and $(1/2) \Gamma^0 \Gamma^5 \Gamma^6$. The scheme of the baryons can be described by a unit cube, the center of which is at the origin of the coordinate axes τ_3, ω_3, f_3 . Proceeding from the two above equations it is possible to determine the perturbational Hamiltonian which describes the weak interaction between the baryons. All processes outlined herein are in principle possible, although some of them (e.g. $\Xi^- \rightarrow \Sigma^- + \pi^0$) cannot be realized owing to the conservation theorems. Other processes (e.g. $\Sigma^+ \rightarrow \Lambda^0 + \beta^+$ + ν) are obviously realizable, but are nevertheless very rare. The scheme advanced herein gives all possible baryon transformations and not even one forbidden transition. The weak interaction transitions which may be expected can proceed from a state with a positive to a state with negative energy and

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A Comment on the Scheme of Baryons

SOV/56-37-2-38/56

vice versa. Hence processes of the type $N + \bar{P} \rightarrow \beta^- + \gamma$ may exist, the probability of which is small as compared to that of strong interaction processes. There are 1 figure and 4 references, 2 of which are Soviet.

ASSOCIATION: Institut fiziki i astronomii Akademii nauk Estoniiskoy SSR (Institute of Physics and Astronomy of the Academy of Sciences of the Estonian SSR).

SUBMITTED: April 16, 1959

Card 4/4

33658

8/058/61/000/012/011/083
A058/A101

24.6610

AUTHOR: Yvglane, Kh.Kh.

TITLE: Baryon decay incident to weak interaction

PERIODICAL: Referativnyy zhurnal. Fizika, no. 12, 1961, 34, abstract 12A443
(Tr. In-ta fiz. i astron. AN EstSSR, 1961, no. 13, 87 .. 111,
English summary)

TEXT: With the aid of mirror transformations of a six-dimensional isotopic space there are plotted isotopic vectors and set up a general wave equation for baryon systems. The "unperturbed" part of this equation describes the stationary states of the baryons, while the "perturbation" Hamiltonian yields the interbaryonic transitions that are due to weak interaction. The "primary" mechanical masses of the baryons differ from each other, this difference in mass being due to the same cause as baryon decay. There is set up an equation for a baryon in an external electromagnetic field. Baryon charge parity is determined, but it is not conserved in decays; only the so-called generalized parities are conserved.

[Abstracter's note: Complete translation]
Card 1/1 X

YYGLANE, Kh. [Uiglane, H.]; KUTUZOVA, G.

Equation for a boson doublet. Zhur.eksp.i teor.fiz. /6 no.3:780-781 Mr '61. (MIRA 1418)

1. Institut fiziki i astronomii Akademii nark Estonskoy SSR.
(Particles (Nuclear physics)) (Differential equations)

IVYGLANE, Kh. [Üiglane, H.]

Marshak invariance and four-fermion interaction. Zbir.eksp. i
teor.fiz. 40 no.3:782-783 Mr '61. (MIRA 14:8)

1. Institut fiziki i astronomii Akademii nauk Estonskoy SSR.
(Atomic nuclei) (Nuclear reactions)

24,6610

S/058/62/000/007/012/068
A061/A101

AUTHORS: Kutuzova, G. B., Vyglane, Kh. Kh.

TITLE: A unified equation for bosons and fermions

PERIODICAL: Referativnyy zhurnal, Fizika, no. 7, 1962, 36, abstract 7A318
("Tr. In-ta fiz. i astron. AN EstSSR", 1961, no. 16, 81 - 89;
English summary)

TEXT: A first-order equation is presented which, depending on the characteristics of wave function transformation, may describe both bosons and fermions. In the case of fermions, the equation obtained decomposes into four Dirac equations, while in the case of bosons it decomposes into two eight-component equations. In the case where the boson momentum differs from zero, it is found impossible simultaneously to determine accurately the boson energy, the spin projection, and the absolute spin value. It is solely possible to determine either the spin projection and the absolute spin value, or the spin projection and the energy. The boson energy operator is found to have also negative eigenvalues. With the operation of charge conjugation it is possible to pass from states with

Card 1/2

A unified equation for bosons and fermions

S/058/62/000, 1007/012/058
A061/A101

negative energy to such with positive energy, but having an opposite sign of charge.

[Abstracter's note: Complete translation]

Card 2/2

S/058/62/000/007/013/068
A061/A101

AUTHOR: Yyglane, Kh. Kh.

TITLE: On some possible generalizations of the group representation concept

PERIODICAL: Referativnyy zhurnal, Fizika, no. 7, 1962, 36, abstract 7A319
("Tr. In-ta fiz. i astron. AN EstSSR", 1961, no. 16, 90 - 105;
English summary)

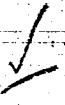
TEXT: A generalized group representation is determined which, in the case of transposed nonsingular matrices of representation, can be reduced to a projective representation. It is shown that the generalized representation of a group of reflections of two-dimensional (or three-dimensional) space yields three-row spin matrices. The projective representation of this group yields two-row spin matrices. Analogously, the generalized representation of a group of reflections of four-dimensional space is described by five or ten-row Kemmer-Daffin β_j matrices. The projective representation of this group is determined by Dirac γ_j matrices. It is shown that generalized and projective representations of a group of space-time reflections do not describe spins greater than unity. It is shown in con-

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On some possible generalizations...

S/058/62/000/007/013/068
A061/A101

clusion that Cailey's non-associative algebra can be considered as a non-associative representation of a group of reflections of three-dimensional space.

[Abstracter's note: Complete translation] 

Card 2/2

8/613/62/000/019/005/006
B108/B186

AUTHORS: Lykhmus, Ya., Yyglane, Kh.

TITLE: Classification of elementary particles according to their interaction

SOURCE: Akademija nauk Estonskoy SSR. Institut fiziki i astronomii. Trudy. no. 19. 1962. Issledovaniya po teoreticheskoy fizike. 113-123

TEXT: On the basis of six-dimensional "internal" space, which for each interaction is split up into the sum of two invariant subspaces, one mathematical formulation of the classification of the elementary particles according to their interactions is given. Classifications according to weak four-fermion, electromagnetic, and strong interactions are established. Each of these excludes the remaining two from exact determination. The interaction Lagrangian is invariant with respect to transformations in the "internal" space. This means that the quantum numbers are conserved in interactions according to which the classification was made. There are 4 tables. The English-language references are:

Card 1/2

Classification of elementary ...

S/613/62/000/019/005/006
B108/B186

M. Gell-Mann. Phys. Rev., 92, 833, 1953; T. Nakano, K. Nishijima, Progr. Theor. Phys., 10, 581, 1953; A. Salam, Nucl. Phys., 2, 173, 1956.

SUBMITTED: December 15, 1961

Card 2/2

YYGLANE, Kh

ÖIGLANE, Harris; HEINOJA, H., red.

[Info. the depths of the microworld] Mikromailma sugga-
vusse. Tallinn, Eesti Riiklik Kirjastus, 1963. 336 p.
(MIRA 17:6)

TYKS, S.R. [Joks, S.R.]

Laryngoscopic and stroboscopic observations on laryngeal conditions
in singers of amateur choirs. Ves. otorin. 21 no.2:83-87 Kr-Ap '59.

1. Iz kafedry bolezney ykha, gorla i nosa (zav. - dots. N.K. Syrda) (MIEI 12:4)
Tarkuskogo gosudarstvennogo universiteta.

(IAHNE, physiol.

in singers, laryngoscopy & roboscopy (Rus.)

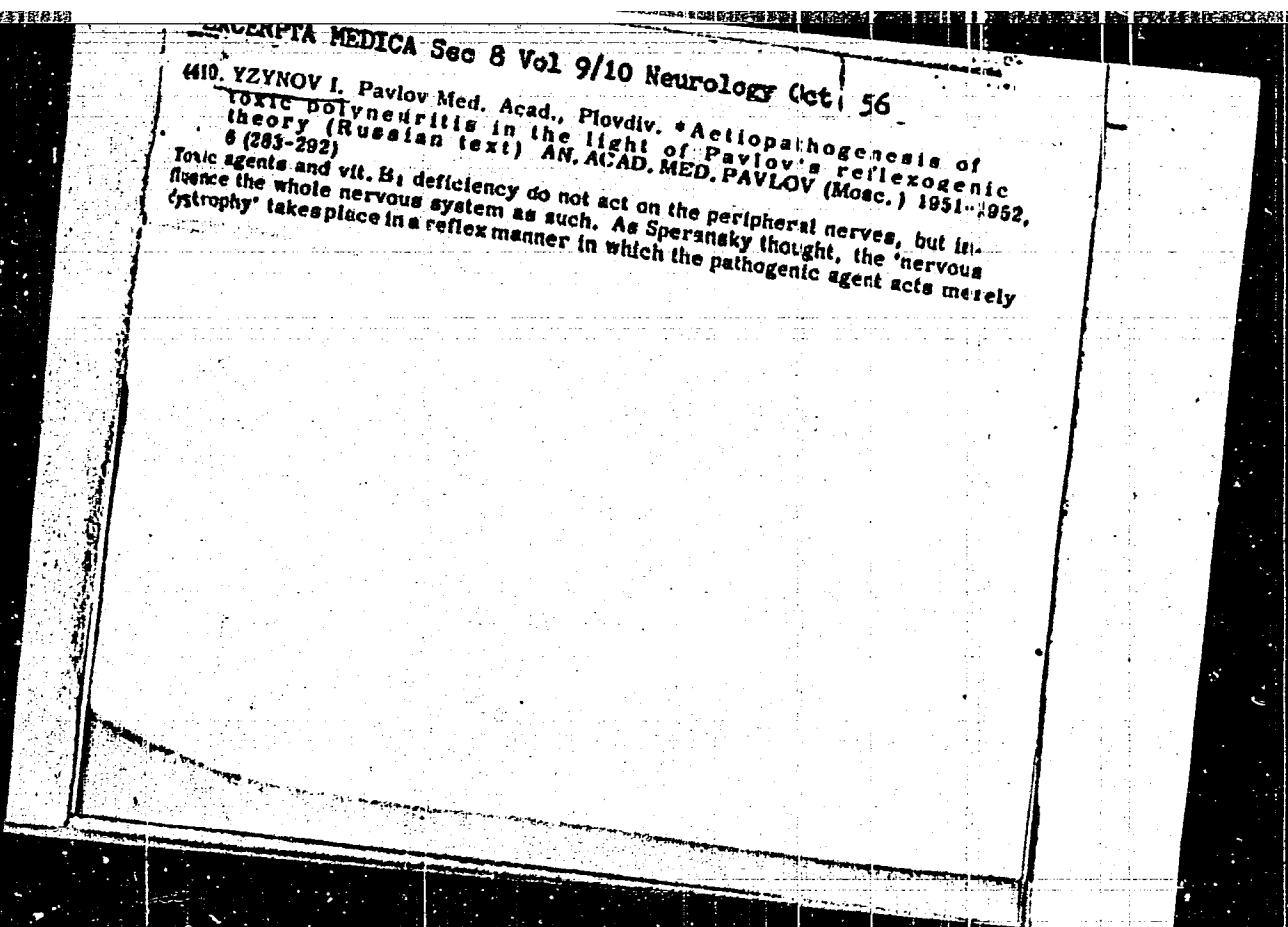
(VOICE,

laryngoscopy & stroboscopy in singers (Rus.)

YYUDU

JÕUDU, Kuata; SILLAMÄA, H., kand. tekhn. nauk, retsenzent;
AEO, L., red.; LAUL, U., tekhn. red.

[Automation] Automaatika. Tallinn, Eesti Riiklik
Kirjastus, 1962. 355 p. (MIRA 16:12)
(Automation)



4410

as a pacemaker in a range of reflex events. An attempt is made to explain dialectically the fact of the selectiveness of the receptive part of the nervous system for toxic agents. (Actually nobody explains the facts from a unitary point of view. Does the submission of all events to one mechanism of reflex function represent a real grasping of the whole from all points of view? Ref.) Kuntic - Belgrade

ZALISHVILI, A.A.

Hygienic characteristics of the nutrition of students of the
Tbilisi printing school. Tudy Tbil. GINOV 6:295-299 '62.

(TBILISI--NUTRITION)

(MIRA 16:2)

ZALISHVILI, A. A., KAPANADZE, P. I., KARTOZIYA, P. T., ROSTOMBEKOVA, N. V.,
TAKTAKISHVILI, S. D., KEMALADZE, A. G., MGALOBLISHVILI, Y. V., SOOTYA, P. I.

"On the study of organized nutrition of various age-related and
industrial groups of population of the Georgian SSR."

report submitted at the 13th All-Union Congress of Hygienists, Epidemiologists
and Infectionists, 1959.

KHMALADZE, A.G.; ZAALISHVILI, A.A.

Method for the determination of fats in milk and milk products.
(MIRA 13:12)
Vop. pit. 19 no. 6:85-86 N-D '60.

1. Iz kafedry gigiyeny (zav. - prof. A.G. Khmaladze) Tbilisakogo
gosudarstvennogo instituta usovershenstvovaniya vrachay.
(MILK—ANALYSIS AND EXAMINATION) (BUTTERFAT)

ZAALESHVILI, I. I.:

ZAALESHVILI, I. I.: "Some problems of the dynamics of bushing-roller chain drive." Min Higher Education USSR. Order of Labor Red Banner Georgian Polytechnic Inst imeni S. M. Kirov. Tbilisi, 1956.
(Dissertation for the Degree of Candidate in Technical Science.)

Knizhnaya Letopis'
No 32, 1956. Moscow.

ZAAALISHVILI, I.I.

SOV/124-58-7-7402

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nr 7, p 12 (USSR)

AUTHOR: Zaalishvili, I.I.

TITLE: Determining the Pressures in the Hinges of a Roller-chain-drive Chain With Allowance for the Inertia Forces (Opredele-niye davlenii v sharnirakh tsepi vtulochnorolikovogo privoda s uchetom inertsionnykh sil)

PERIODICAL: Tr. Gruz. politekhn. in-ta, 1957, Nr 2(50), pp 149-153

ABSTRACT: Allowance being made for the inertia forces, formulae are evolved for the calculation of the pressures exerted by a pin and sprocket tooth on the bushing, and the results of calculations made with these formulae are compared with results obtained from formulae not allowing for the influence of the inertia forces.

S.G. Kislitayn

1. Drive sprockets--Analysis

Card 1/1

ZAA LISHLVILI, Irina Mikhaylovna,

Academic degree of Doctor of Med Sci based on her defense,
27 January 1953, in the Council of the Tblisi State Med
Inst, of her dissertation entitled: "Research on peri-
pheric blood circulation in hypertonic disease".

Academic degree and/or title: Doctor of Sciences

SO: Decisions of VAK, List no 7, 26, Mar 55, Byulleten'
MVO SSSR, No. 14, July Moscow pp 4-22, Uncl.
JPRS/NY-429

ZAAALISHVILI, I.M. (Tbilisi)

Vencus pressure in experimental inflammation. Pat. fiziol. i eksp.
terap. 5 no.2:32-35 Mr-Ap '61. (MIRA 14:5)

1. Iz kafedry patologicheskoy fiziologii (zav. - I.M.Zaalishvili)
Tbilisskogo mediteinskogo instituta.
(INFLAMMATION) (BLOOD PRESSURE)

ZAALISHVILI, I. M.

USSR / Human and Animal Physiology. Blood Circulation.

Abs Jour: Ref Zhur-Biol., No 9, 1958, 41326.

Author : Zaalishvili, I. M.

Inst : Institute of Cardiology AS. GSSR. Tbilisi.

Title : Data of Studies of the Peripheral Circulation in Hypertension.

Orig Pub: V sb.: Stenogr. otchet nauchn. sessii in-ta, Kardiol. AN GruzSSR s uchastyem in-ta fiziol. AN USSR Tbilisi, AN GruzSSR, 1956, 110-118 (Stenographic transcript of the Scientific Session of the Institute of Cardiology AS. GSSR with the participation of the Institute of Physiology AN USSR)

Abstract: A spastic-vagotonic syndrome was noted in patients

Card 1/3

USSR / Human and Animal Physiology. Blood Circulation. T

Abs Jour: Ref Zhur-Biol., No 9, 1958, 41:26.

Abstract: with hypertension (H) with the aid of capillaroscopic investigations. During the early stages of H capillary anastomoses were observed in 20-30% of cases. The length of the capillary loop is increased (at first in its venous portion) and the distance between the crests of the loops is greater. Capillary hemorrhages were noted. The pressure (P) in the arteries of the fingers was at the upper limit of normal during the initial stages of H, in the later stages it reached from 150-230mm of mercury. The author thinks that the increased tonus of medium caliber arteries decreases in the later stages of H. The effect of emotion on P in the arteries of the fingers was noted. P in the finger arteries increased or decreased as the patients' position was passively changed from the

Card 2/3

67

ZAAALISHVILLI, I. M.

Importance of blood vessels of the muscular type in regulating
blood pressure in health and in pathology. Trudy Inst. klin.
i eksper. kard. AN Gruz. SSR 8:349-352 '63. (MERA 17:1)

I. Kafedra patofiziologii Meditsinskogo instituta, Tbilisi.

ZALISHVILI, M.M.; SHRAYBER, F.O.

A device for automatic application of test solution to the chromatographic paper. Biokhimiia 28 no.1:9-12 Ja-F '63.
(MIRA 16:4)

1. Institut fiziologii Akademii nauk Gruzinskoy SSR, Tbilisi.
(PAPER CHROMATOGRAPHY)

ZAAALISHVILI, M.M.; KLEYN, Ye.E.

Microcclerimetric determination of histamine. Seob.AH Gruz.SSR 9
no.3:167-171 '48. (MLM 9:7)

I.Akademiya nauk Gruzinsskoy SSR, Institut fiziology imeni
I.S.Beritashvili, Tbilisi. Predstavlene chlenen-korrasjudentem
Akademii P.A.Konetziani.
(Histamine)

CR ZALISHVILI, M.M.

108

Physicochemical and biochemical characteristics of myogen A. M. M. Zalishvili (Georgian Acad. Sci., Tbilisi, U.S.S.R.). *Biochimya* 16, 229-33 (1951).—The aldolase activity of myogen A (I) was higher than the activity of the initial muscle ext. After recryst., the aldolase activity of I increased. I contained 0.9% free amino groups, 1.8% carboxyl groups, no free sulfhydryl groups, 1.96% tryptophan, and 7.77% lysine. The ultraviolet absorption max. was at 2780 Å. Numb rabbit muscle, produced by a faradic current nerve stimulation, yielded cryst. I possessing the identical properties of I obtained from normal rabbit muscle. H. Priestley

ZALISHVILI, M.M.

Monomolecular layer of myogen A. Biokhimiia, Moskva 16 no. 4:321-327
(ChML 21:1)
July-Aug 51

1. Institute of Physiology, Academy of Sciences ASRR, Tbilisi.

ZAALISHVILI, M.M.; MIKADZE, G.V.

Role of actin in muscle tissues and some problems in the theory
of muscle contraction. Biokhimia 24 no.4:612-624 J1-Ag '59.
(MIRA 12:11)

1. Institut fisiologii Akademii nauk Gruzinskoy SSR, Tbilisi.
(MUSCLE PROTEINS metab)

ZALISHVILI, M.M.; SHRAYBMAN, F.O.; YEGYAZAROVA, A.R.

Apparatus with automatic control for the determination of the diffusion coefficient. Biofisika 5 no.1:69-75 '60. (MIRA 13:6)

1. Institut fiziologii AN Gruzinskoy SSR, Tbilisi.
(TECHNOLOGY RADIOLOGIC equip. & supply)

ZAALESHVILI, M.M.) SHRAYEMAN, F.O.

Apparatus for automatic application of spots to the chromatographic paper. Biokhimiia 25 no. 3:570-572 My-Je '60. (MIRA 14:4)

1. Institute of Physiology, Academy of Sciences of Georgian S.S.R.,
Tbilisi.

(PAPER CHROMATOGRAPHY)

ZAALESHVILI, M.M.

Interaction of myosin and actomyosin with adenosinetriphosphoric acid. Biokhimiia 25 no.5:912-919.S-0 '60. (MIHA 14:1)

1. Institute of Physiology, Academy of Sciences of the Georgian S.S.R., Tbilisi.
(ADENOSINETRIPHOSPHORIC ACID) (MYOSIN)

ZAAKISHVILI, M. M., and DZHIBLADZE, S. V. (USSR)

"Some Data of Contractile Proteins of Tonic and Tetanic Muscles."

Report presented at the 5th International Biochemistry Congress,
Moscow, 10-16 Aug 1961

ZAALISHVILI, M.M.

Interaction of myosins A and B with adenosine triphosphate and
some problems in the theory of muscular contraction. Trudy Inst.
fiziol. AN Gruz. SSR 12:175-192 '61. (MIRA 15:2)
(MYOSIN) (ADENOSINE TRIPHOSPHATE)
(MUSCLE MOTILITY)

ZHALISHVILI, M.M.; SHRAYBMAN, F.O.

Device for the measurement of the initial rate of an enzymatic reaction. Biokhimiia 27 no.1:72-76 Ja-F '62. (MIRA 15:5)

1. Institute of Physiology, Academy of Sciences of Georgian S.S.R.,
Tbilisi.
(ENZYMES)

ZAAKISHVILLI, M.M.; SURGULADZE, T.T.; YEGIAZAROV, A.R.; COGORISHVILI,
Dzh.A.

Studying the interrelation of myosin A and myosin B with
adenosine triphosphate by the method of electrophoresis.
Soob. AN Gruz. SSR. 30 no.1:29-36 Ja '63. (MIR 17:1)

1. Institut fiziologii AN Gruzinskoy SSR, Tbilisi.
Predstavleno akademikom P.A. Kometiani.

ZAAVISHVILI, M.M.; DZHIBLAZIE, S.V.

Nature of the contractile proteins of tonic and tetanic skeletal muscles. Soob. AN Gruz. SSR 31 no.1:53-60 Jl '63.
(MIRA 17:7)

1. Predstavleno akademikom P.A. Kometiani.

Zaalishvili, M.M.; Mikadze, G.V.

Some problems of mechanical chemistry of the smooth muscle.
Biochimia 29 no.5:801-811 J1-Ag '64. (MJRA 18:11)

1. Institut fisiologii AN GruzSSR, Tbilisi.

L 08289-67 EWT(1) RO

ACC NR: AP7000434 SOURCE CODE: UR/0251/66/044/002/0311/0316

AUTHOR: Zaalishvili, M. M.; Dzhibladze, S. V.23
BORG: Institute of Physiology, AN Georgian SSR (Institut fiziologii Akademii nauk Gruzinskoy SSR)TITLE: Cholinesterase activity in myosinSOURCE: AN GruzSSR. Soobshcheniya, v. 44, no. 2, 1966, 311-316

TOPIC TAGS: cholinesterase, enzyme, adsorption, myosin

ABSTRACT: Rabbit-muscle myosin A was obtained according to a standard method, while the frog muscle myosin was prepared as follows: frog sartorius muscles were frozen and then ground and the homogenate extracted with Straub's solution at pH 6.5. The extract was further separated by centrifugation and myosin precipitated with 15 volumes of water cooled to 0°C. Thereafter myosin was again precipitated and finally suspended in the equivalent of a 1.2-M KCL solution. Total nitrogen was determined by the Kjeldahl method and ATP by Lyubimova's method, while cholinesterase activity was assayed according to Varga's method. To evaluate the effect of temperature on the cholinesterase activity of myosin, the

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L 08289-67
ACC NR: AP7000434

myosin suspensions were exposed to varying temperatures for periods of 5, 10, 15, and 20 min. To separate the cholinesterase fraction from myosin, the 0.6-M KCL myosin suspension was heated to 50C for 10 min, cooled to 0C for several hours, and the precipitated fraction removed by centrifugation. At this stage, cholinesterase activity was determined for the stock suspension and the supernatant fraction. Myosin heated to 50C loses ATP-ase activity, but cholinesterase activity remains unaltered. Due to thermal denaturation of proteins, cholinesterase myosin activity is shifted to the supernatant fraction. It is therefore concluded that the "false" cholinesterase activity is due to enzyme adsorption by myosin.

Orig. art. has: 2 figures and 1 table.

SUB CODE: 06 / SUBM DATE: 24Nov65 / ORIG REF: 008 / OTH REF: 014 /

Card 2/2 LS

ZAALISHVILI, Sh. D.

Moscow

Laboratory of the Technology of Inorganic Substances, Chemico-
Technological Institute imeni D.I. Mendeleyev, (-1939-).

"The Solubility of Carbon Dioxide from Mixtures with Hydrogen and Nitrogen
in Water under Pressure."

Zhur. Fiz. Khim., Vol. 14, No. 3, 1940.

ZAALESHVILI, SH. D.

Hydrocarbons

Modified theory of corresponding states and its application to pure hydrocarbons.
Part I. Zhur.fiz.khim., 16, No. 6, 1952.

Monthly List of Russian Accessions, Library of Congress, November 1952. Unclassified.

ZAMASHVILI, Sh. D.

USSR / Chemistry - Hydrocarbons

JUN 52

"Mechanized Theorem of Corresponding States and its Verification on the Example of Pure Hydrocarbons". Part I, "Sh. D. Zamashvili, Gor'kiy Polytech Inst. Izdat. N.A. Zhdanov

Journal "Zh. Khim. No 6, pp 882-891

The new method of computing the dependence $P - V$ proposed by the author has been found applicable from methane to pentane inclusively. There are reasons to believe that it will be valid for hydrocarbons as well. Under the circumstances it will be possible to predict for every

gas of the paraffin series the vol changes within a wide range of pressures and temps with a precision sufficient for engineering calcs. By introducing 3 new constants which by definition make the value of the crit coeff unchangeable, coincidence of hydrocarbon curves (cf. Brown, Sonnert, and Barth, Ind Eng Chem, Vol XXXV, 513, 1932) up to 700 atm was achieved.

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ZAAZISHVILI, SH. D.

Jul 52

LESN/Chemistry - Gas Mixture

"Modified Theorem of Corresponding States for Gas Compounds and Its Verification for Hydrocarbons, II.", Sh. D. ZaaZishvili, Gor'kiy Polytech Inst. ineni A. A. Zhdanov

"Zhur Fiz Khim, Vol 26, No 7, pp 970-976

Most previous works on the theorem of corresponding states has been confined to pure substances. Key's empirical method may be applied to gas mixts with the use of exptl data for p-V-T and with the help of Lenn and pressure. Author applied his own method to gas solns. The principle of corresponding states and the values of pseudocritical constants of gas compns are utilized by him in the same manner as for pure gases. Sage's and Lacy's exptl data are called on to verify the author's work with these mixts: methane-n.-butane, methane-n.-pentane. Verification of the method on 15 gas mixts permitted these conclusions: the system methane-ethane gives unsatisfactory results; the system ethane-carbon dioxide gives excellent results (forall compns); the system methane-n.-pentane, with large concns of methane, behaves in accordance with the principle. The author states that the method of calcg p-V-T data for pure gases can be applied to the gas mixts, methane-hydrocarbon with large

(2)

248T8

dilutions of methane and at sufficiently high temps (300°-500° K), to these mixtures under high pressures.

(3)

248T8

ZAHLISHVILI, SH. D.

AID P - 1275

Subject : USSR/Chemistry

Card 1/1 Pub. 119 - 4/5

Author : Zaalistvili, Sh. D. (Gor"kiy)

Title : Corresponding states

Periodical : Usp. khim., 23, no. 7, 867-875, 1954

Abstract : Review of literature based principally on non-Russian sources. Discussion of the application of the theorem of corresponding states for determining the properties of gases at high pressures. 45 references (6 Russian: 1882-1952).

Institution : None

Submitted : No date

ZAALIASHVILI, SH.D.

ADD P - 3163

Subject : USSR/Chemistry

Card 1/1 Pub. 119 - 5/7

Author : Zealishvili, Sh. D. (Gor'kiy)

Title : The second virial coefficient for pure gases

Periodical : Usp. khim., 6, 759-778, 1955

Abstract : The literature dealing with second virial coefficients for individual gases is reviewed. Attention is drawn to the need for precise low-pressure V data for gases. This information is of use in establishing laws of molecular interaction in gases. Fifty tables, 80 references, 2 Russian (1952).

Institution : None

Submitted : No date

Zealishvili, Sh.D.

Category : USSR/Atomic and Molecular Physics - Gases

D-7

Abs Jour : Ref Zhur - Fizika, No 3, 1957, No 6378

Author : Zealishvili, Sh.D.

Title : Second Virial Coefficients for Gas Solutions

Orig Pub : Zh. fiz. khimii, 1956, 30, No 8, 1891-1895

Abstract : Data on the second virial coefficient for gas solutions, scattered in various sources, are collected together.

Order : 1/1

5 (4)
AUTHOR:Zaalishvili, Sh. D.S/076/60/034/01/016/044
B008/B014

TITLE:

Application of Hermite's Interpolation Polynomial for the
Description of Physicochemical Data

PERIODICAL:

Zhurnal fizicheskoy khimii, 1960, Vol 34, Nr 1, pp 102-104
(USSR)

ABSTRACT:

In this paper the author attempted to use the Hermitian polynomial for calculating the critical coefficient $P_c V_c / R T_c$ of a pure substance. A polynomial was set up by the methods described in reference 4. Equation (5) was derived for the critical isotherm. Herefrom it was possible to derive a straightforward formula for the critical coefficient (6). In order to verify equation (6), table 1 was compiled in accordance with experimental data on the critical parameters (Ref 6) and the second virial coefficients of gases (Ref 7). The same problem was solved in consideration of the third virial coefficient: formula (7). The values required for the verification of this formula were calculated from the equation of state derived by Meyers (Ref 5). The experimental and calculated values of the critical coefficient were in close agreement in all cases.

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Application of Hermite's Interpolation
Polynomial for the Description of Physico-
chemical Data.

S/076/60/034/01/016/044
B008/B014

The author thanks I. R. Krichevskiy for his suggestion to solve
the afore-mentioned problem with the help of the Hermitian
polynomials. There are 2 tables and 7 references, 3 of which
are Soviet.

ASSOCIATION:

Gor'kovskiy politekhnicheskiy institut im. A. A. Zhdanova
(Gor'kiy Polytechnic Institute imeni A. A. Zhdanov)

SUBMITTED:

April 13, 1960

Card 2/2

ZAAALISHVILLI, Sh.D.; KOLYSKO, L.E.

Second virial coefficient for vapors and their mixtures. Part 1:
system diethyl ether - acetone. Zhur. fiz. khim. 34 no. 11:2596-
2600 N '60. (MIRA 14:1)

1. Gor'kovskiy politekhnicheskiy institut im. A.A. Zhdanova.
(Ether) (Acetone) (Equation of state)

S/076/61/035/011/011/013
B101/B110

AUTHORS: Zaalishvili, Sh. D., and Kolysko, L. E.

TITLE: The second virial coefficient of vapors and their mixtures.
II. The system acetone - chloroform

PERIODICAL: Zhurnal fizicheskoy khimii, v. 35, no. 11, 1961, 2613 - 2615

TEXT: Since knowledge of the second virial coefficient is necessary for the exact determination of data for the equilibrium between liquid and vapor, the authors investigated the system acetone - chloroform which is typical of systems with maximum boiling point. Purified CHCl_3 and $(\text{CH}_3)_2\text{CO}$ were vaporized in a chamber through which dry nitrogen was blown. The chamber was designed by the Laboratoriya vysokikh davleniy GIAP (Laboratory of High Pressures of the GIAP) (I. R. Krichevskiy, Yu. V. Tsekhaneskaya, Zh. fiz. khimii, 10, 2315, 1956). The compressibility of mixtures containing 14.13; 32.40; 44.50; and 74.59 mole% of acetone was measured, and the second virial coefficient B_m determined. Experimental data follow the equation $B_m = B_{11}N_1^2 + 2B_{12}N_1N_2 + B_{22}N_2^2$ (1), where B_{11} and N_1

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S/076/61/035/011/013
B101/B110

PAGE 1

The second virial coefficient...

are the second virial coefficient and the molar part of acetone, respectively. B_{22} and N_2 are the second virial coefficient and the molar part of chloroform, respectively. B_{12} is the second virial coefficient of the mixture. The maximum deviation of experimental data from those calculated according to this equation is 3%. The B_m values were tabulated:

$t, {}^\circ\text{C}$	$-B_m, \text{cm}^3/\text{mole}$						$-B_{12}$
	0	20	40	60	80	100	
60	910	1275	1500	1585	1530	1330	2005
70	855	1010	1115	1170	1180	1140	1300
80	775	855	920	965	995	1010	995
90	710	785	840	865	865	845	930

The volatility f_1 , f_2 of acetone and chloroform, respectively, at 55°C was calculated by means of the equations:

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S/076/61/035/011/011/013
E101/B110

The second virial coefficient...

$RT \ln f_1 = RT \ln PN_1 + [B_{11} - (B_{11} - 2B_{12} + B_{22})N_2^2]P;$
 $RT \ln f_2 = RT \ln PN_2 + [B_{22} - (B_{11} - 2B_{12} + B_{22})N_1^2]P$ (3), and compared
with the partial pressure (data by H. Röck, W. Schröder, Z. phys. Chem.,
11, 41, 1957) (Table 3). Considerable deviations between partial pressure
P and volatility f show that the second virial coefficients should be
considered when investigating the equilibrium vapor - liquid. The authors
thank I. R. Krichevskiy for advice. M. I. Shumilina and O. S.
D'yachkovskaya assisted in the experiments. There are 1 figure, 3 tables,
and 7 references: 3 Soviet and 4 non-Soviet. The reference to the
English-language publication reads as follows: J. Timmermans, Physico-
chemical constants of pure organic compounds, 1950.

ASSOCIATION: Gor'kovskiy politekhnicheskiy institut im. A. A. Zhdanova
(Gor'kiy Polytechnic Institute imeni A. A. Zhdanov)

SUBMITTED: October 11, 1960

Card 3/43

MADYANOV, A.M., kand. tekhn. nauk, dots.; TIKHONOV, G.F., kand. tekhn. nauk, dots., ovt. red.; ZAALISHVILI, Sh.D., doktor khim. nauk, prof., retsenzent; ASTROV, Ye.I., kand. tekhn. nauk, dots., retsenzent; KOZYULINA, R.M., red.

[Principles of the theory of metallurgical processes;
manual for students of the department of metallurgy]
Osnovy teorii metallurgicheskikh protsessov; uchebnik
posobie dlia studentov metallurgicheskogo fakul'teta.
Gor'kii. Pt.2. 1962. 112 p. (MIRA 17:3)

1. Gorkyi. Politekhnicheskiy institut. Kafedra litaynogo
proizvodstva.

ZAAALISHVILI, Sh.D.; KOLYSKO, L.E.; Prinimala uchastiyu: SHUMILKINA, M.I.

Second virial coefficient of vapors and their mixtures. Part 3.
The system diethyl ether - methyl iodide. Zhur. fiz. khim. 36
no.4:846-848 Ap '62. (MIRA 15:6)

1. Gor'kovskiy politekhnicheskiy institut imeni A.A.Zhdanova.
(Ethyl ether) (Methane) (Virial coefficients)